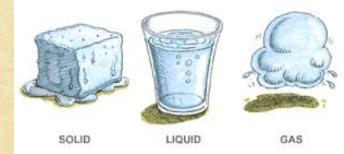
What Are the States of Matter?

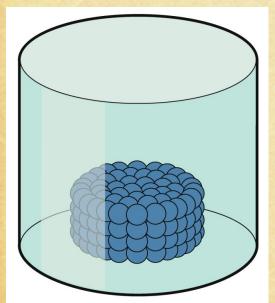
Chapter 10 Lesson 2 Part 1

ByDesign Science, Level 4 By Allyssa Sharpe

 An important property of matter is its state at room temperature.

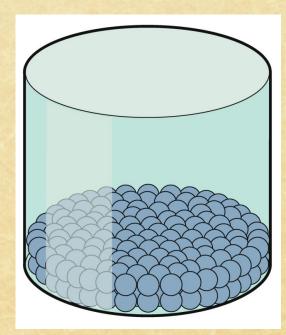


- Solid, liquid, and gas are three states of matter that are common on Earth.
- The fact that matter can exist in different forms illustrates the design and care by which God creates.
- You can tell if matter is a solid, liquid, or gas by analyzing its shape and volume.



- A solid has a definite volume and a definite shape.
- An ice cube is a solid.
- Look around and see if you can find other examples of solids.

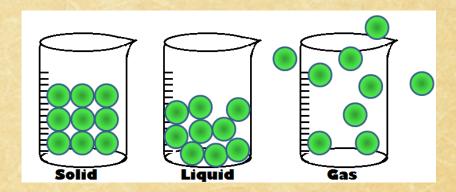
- A liquid has a definite volume, but it has an indefinite shape.
- A liquid can change shape if you place it into a different container.
- Water is a liquid at room temperature.



- If you pour some water into a glass, it takes the shape of the glass.
- Suppose you pour the same water into a vase.
- Its volume is the same, but its shape changes.

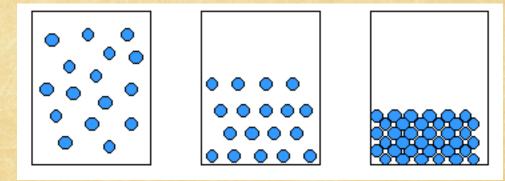


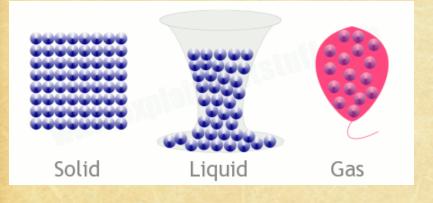
- A gas has both an indefinite volume and an indefinite shape.
- A gas takes the shape of its container.
- Air is made up of different gases.



- If you blow up a balloon, the air takes the shape of the balloon.
- When you let the air out of a balloon, the gases in the air spread all around.
- A gas has an indefinite shape and always spreads out to fill any available volume.

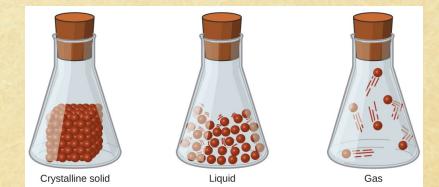
- You have learned that all matter is made of tiny particles called atoms.
- The atoms in matter have different arrangements in each different state of matter.
- Atoms also move differently in the different states of matter.



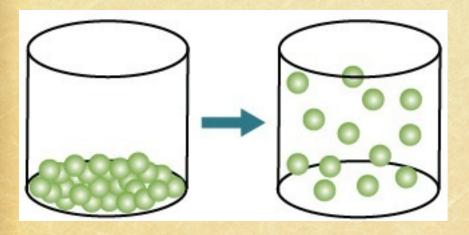


- A solid keeps a certain shape because its atoms are packed closely together.
- The atoms in matter are always moving.
- In a solid, the atoms vibrate, or move back and forth.
- They do not exchange places with one another.

 Atoms in a liquid are farther apart than those in a solid, and they vibrate faster.

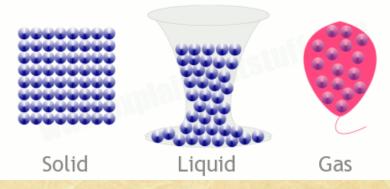


- The atoms are not packed together, so they can slide around one another.
- Therefore you can pour a liquid and why a liquid can change its shape.



- The atoms in a gas are not packed together strongly.
- They are farther apart than those in a liquid.
- Therefore it is easier to move through the air than to push through the water in a swimming pool.

 Gas particles move around much faster than atoms of solids or liquids.



- They move around freely, and bounce off one another.
- So, they can spread out and fill any available space.

